

Flavonoid oxidation in plants: from biochemical properties to physiological functions

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Flavonoids protect plants against various biotic and abiotic stresses, and their occurrence in human diet participates in preventing degenerative diseases. Many of the biological roles of flavonoids are attributed to their potential cytotoxicity and antioxidant abilities. Flavonoid oxidation contributes to these chemical and biological properties and can lead to the formation of brown pigments in plant tissues as well as plant-derived foods and beverages. Flavonoid oxidation in planta is mainly catalyzed by polyphenol oxidases (catechol oxidases and laccases) and peroxidases. These activities are induced during seed and plant development, and by environmental stresses such as pathogen attacks. Their complex mode of action is regulated at several levels, involving transcriptional to post-translational mechanisms together with the differential subcellular compartmentalization of enzymes and substrates.

Flavonoids

Flavonoids are widely distributed plant secondary metabolites resulting from the addition of malonyl CoA the phenylpropanoid molecule coumaroyl CoA to (Figure 1a) [1–6]. These polyphenolic compounds are characterized by two aromatic cycles (A- and B- rings) linked by a heterocycle (C-ring) (Figure 1b). They are classified according to the oxidation degree of the C-ring, and include flavonols, anthocyanins and flavan-3-ols. These molecules can undergo modifications of their aromatic cycles, including hydroxylations, methylations, glycosylations, acylations or prenylations, which account for the diversity within a compound class [5,7]. The condensation of flavan-3-ols leads to the formation of proanthocyanidins (PAs), also called condensed tannins [4,8]. Phlobaphenes are reddish-brown pigments that are present in several plant tissues including the maize pericarp. They are poorly defined polymers hypothesized to result from non-enzymatic oxidation of colorless flavan-4-ols [9].

Flavonoids exhibit central functions in various aspects of plant life related to interactions with the environment [2,3,5,10]. For instance, they protect the plant against ultraviolet radiations. They also have antimicrobial properties and act as a deterrent for herbivores by limiting assimilation of dietary proteins and inhibiting digestive enzymes [4,8,11]. Flavonoids also play important roles in human health through consumption of plant-derived foods, by preventing degenerative diseases associated with oxidative stress [12,13].

Biological effects of flavonoids are linked to their potential cytotoxicity and their capacity to interact with enzymes through protein complexation. Furthermore, flavonoids act as scavengers of free radicals such as reactive oxygen species (ROS), and also prevent their formation by chelating metals [11,12,14,15]. In this context, it is relevant to consider how the oxidation of flavonoids affects their biological properties [16]. Interestingly, kaempferol and quercetin polymers catalyzed by polyphenol oxidases have been shown to have a stronger scavenging effect on ROS than their respective monomers in human cell lines [17]. Quinones are powerful antibiotics [18], which, apart from having the same tanning properties as their reduced counterpart, can also alkylate proteins. Quinones generated by polyphenol oxidases have been shown to limit the accessibility of alkylatable dietary proteins to plant-feeding insects, thus causing their starvation [19].

This article reviews our current understanding of flavonoid oxidation and its impact on plant physiology. The relationships between flavonoid oxidation and browning are analyzed. Moreover, the different oxidases (polyphenol oxidases and peroxidases) involved in these processes, together with the different levels of regulation that control browning reactions in plants are described. Finally, several examples illustrating the relationships between flavonoid oxidation and plant defense mechanisms are presented.

In planta flavonoid oxidation: a browning process

Flavonoids, particularly *ortho*-diphenols (*o*-diphenols), can be oxidized to their corresponding semiquinones and quinones by oxidases such as polyphenol oxidases (PPO) and peroxidases (POD) (Figure 2a). Semiquinones and quinones are highly reactive species that undergo further non-enzymatic reactions. They can spontaneously react with phenols, amino acids or proteins, yielding a complex mixture of brown products [20–22] (Figure 3a). Non-enzymatic

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Review



Figure 1. Structure, biosynthesis and oxidation of flavonoids. (a) Simplified schematic of the flavonoid pathway. The main classes of end-products are presented, and their molecular structure illustrated by one example for each class. (b) The structure of the flavonol quercetin is given as an example of carbon numbering. Important features influencing antioxidant potential are the di-hydroxylated B-ring, unsaturation at the C-ring and a 4-oxo function at the C-ring [13].

oxidation of flavonoids, such as autoxidation and chemical oxidation, can also lead to the formation of quinoidal compounds [21,23]. Through coupled oxidation reactions, quinones can oxidize other polyphenols that cannot be directly oxidized by the enzymes, thereby forming secondary quinones, which, in turn, contribute to the formation of heterogeneous polymers responsible for the browning reaction [24] (Figure 3b). The composition of these brown polymeric species is extremely difficult to characterize *in vivo* [20,22].

In Arabidopsis, seed coat browning is caused by the oxidation of epicatechin and soluble PAs by the TRANSPARENT TESTA 10 (TT10) laccase during the developmentally determined desiccation phase [6,25,26](Figure 4). Seeds of the tt10 null mutant are yellow at harvest, but slowly turn brown during postharvest storage until they eventually resemble wild-type seeds. They show an increase in soluble (non- or poorly oxidized) PAs and a higher ratio of quercetin-rhamnoside (QR) monomers to dimers. The contribution of QR dimers to browning is unclear. A similar enzyme could be responsible for the oxidative polymerization of flavan-4-ols leading to reddish-brown phlobaphenes in maize pericarp. Postharvest browning is also observed in pinto bean lines (Phaseolus vulgaris var. Pinto) [27]. During seed development, molecules of catechin react with kaempferol to yield heterodimers that accumulate in the seed coat. Interestingly, seed coat darkening with aging can be correlated with an increase in the amount of dimers. The browning of the peel of litchi fruits has been correlated with the rapid degradation of red anthocyanin pigments [28]. The authors suggest that to produce brown pigments, anthocyanins must become accessible to oxidation by a PPO or a POD after enzymatic removal of their sugar moieties. Other authors have shown that the major litchi anthocyanin, cyanidin 3-rutinoside, is oxidized by litchi POD and by the epicatechin quinone generated by PPO oxidation of epicatechin, through a coupled oxidation process (E. Le Roux, PhD thesis, University of Aix-Marseille, 1999) (Figure 3b).

The enzymatic oxidation of polyphenols, particularly flavonoids, also occurs during food processing of plant

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Figure 2. Three key enzymes for flavonoid oxidation in plants. (a) Enzymatic reactions for polyphenol oxidases (catechol oxidase and laccase) and peroxidases. (b) The main structural features of plant polyphenol oxidases and peroxidases are presented and compared. Poplar (*Populus* sp.) laccase (*Pe* LAC3 [33]), catechol oxidase (*Ptd* PPO1 [78]), and peroxidase (*Pt* PXP 3-4 [39]) are considered as examples. The subcellular localization was predicted by the SignalP server. The catalytic and N-glycosylation sites are PROSITE patterns. P1 and P2 stand for the active and heme-binding sites, respectively. Abbreviation: aa, amino acids. (c) Catalytic mechanisms for the polyphenol oxidases of the catechol oxidase and laccase types [36,37,48]. CuA, CuB and T are copper-binding sites.

material as well as during storage, when cell integrity is affected. The resulting browning is one of the main causes of quality loss [20,29]. Several techniques have been developed to inhibit browning, including physical methods (e.g.

heat, modified atmosphere) and chemical inhibitors (e.g. ascorbic acid, halide ions, carboxylic acids) that can affect either the enzyme, the substrate or the product [20,22,30]. Genetic engineering also offers alternatives for producing

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Figure 3. Mechanisms of flavonoid oxidation leading to brown polymers. (a) Enzymatic interactions leading to browning [20,48]. (b) Coupled oxidation of flavonoids [37]. As an example, the initial reaction is catalyzed here by a laccase. Broken arrows indicate unknown oxidation mechanisms. Abbreviation: R, radical.

fruits and vegetables with increased resistance to enzymatic browning. For instance, potatoes expressing sense or antisense RNA from tomato PPO show less enzymatic browning [31].

Three key enzymes

Three major enzymes, laccase (EC 1.10.3.2), catechol oxidase (EC 1.10.3.1) and peroxidase (EC 1.11.1.7), are known to be involved in flavonoid oxidation (Figure 2). Laccase and catechol oxidase belong to the PPO category.

Laccases (LAC) are *o*-diphenol and para(p)-diphenol: dioxygen oxidoreductases belonging to a group of enzymes called blue copper oxidases, which includes, among others, ascorbate oxidase and ceruloplasmin (Figure 2a). These enzymes are multi-copper glycoproteins that catalyze the oxidation of diphenolic substrates in the presence of molecular oxygen [32,33]. The catalytic sites are composed of four histidine-rich copper-binding domains (Figure 2b,c).

Catechol oxidases (CO) refer to metalloenzymes catalyzing the oxidation of *o*-diphenols to the corresponding *o*-quinones (catecholase or diphenolase activity) (Figure 2a). Generally, COs also catalyze the oxidation of monophenols to *o*-diphenols (cresolase or monophenolase activity, EC.1.14.18.1) [34] (Figure 3a). They are moderately glycosylated copper-binding proteins using molecular oxygen as a cofactor and exhibiting two copper-binding domains (Figure 2b,c). Together with laccases, they are present in eukaryotes and prokaryotes [32,34–37].

Peroxidases (POD) are hemoproteins that catalyze the oxidation of phenolic substrates through the associated reduction of hydrogen peroxide in the peroxidative cycle (Figure 2a). They are able to produce ROS such as a

superoxide anion $(O_2^{-\bullet})$ or a hydroxyl radical (OH^{\bullet}) through the hydroxylic cycle [38,39]. Plants possess only two classes of PODs, which differ according to the subcellular localization of their members (see below).

A major issue encountered when analyzing browning processes is identifying which of the three oxidoreductase(s) is involved. Enzymatic assays can be performed with the purified enzyme in the presence of selective substrates and inhibitors. However, not all plants possess the three types of enzymes. For instance, *Arabidopsis* PPOs are represented only by laccases (no gene encoding catechol oxidase enzymes has been found in the genome) [26].

The different levels of control

From gene expression to protein activation

Most of the plant PPOs and PODs belong to multigenic families, whose members can exhibit functional redundancy when simultaneously expressed in the same tissue [26,33,40–42]. In Arabidopsis, the TT10 laccase appears to be the only member involved in flavonoid oxidation in seeds given that neither of the 16 other laccases, nor any POD, can compensate for the loss of its activity in the tt10 mutant. TT10 is mainly expressed in the seed coat (testa). The TT10 laccase colocalizes with flavonoids and is already present in tissues long before testa browning is observed [26]. Multigenic families of flavonoid oxidases, such as LACs, COs and PODs, have been maintained during evolution, suggesting that each enzyme is needed to carry out different specific functions in planta. In tomato, the seven members of the CO gene family are differentially induced or downregulated in various tissues and by various types of environmental stresses, suggesting that they have

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Figure 4. Seed coat pigmentation in *Arabidopsis*: illustration of a browning process. (a) Photographs showing the appearance of a brown pigmentation in the testa of the wild-type genotype during seed desiccation. The brown pigment is absent from the *transparent testa 10 (tt10)* mutant defective in a laccase enzyme. (b) Mutant *tt10* seeds slowly become brown after harvest and eventually resemble wild-type seeds. Scale bar = 550 µm. (c) Schematic drawing indicating the occurrence of brown pigmentation during *Arabidopsis* seed development. Abbreviation: DAF, days after flowering.

different functions [43]. In apricot, although the amount of PPO is stable during fruit development, a peak of PPO activity is recorded at the breaker stage when no mRNA is present. This observation suggests that the protein is stable and that a post-translational control regulating PPO activity occurs [44]. Most plant PPOs are present in a latent form that requires activation [45]. For example, in apple, latent CO can be activated *in vitro* by the addition of SDS. This activation process is caused by a reorganization of the protein tertiary structure, which modifies the access of substrates to the active site of the enzyme [46,47]. *In planta*, endogenous proteases are involved in the activation of latent COs, by cleaving a 15–20 kDa C-terminus fragment, yielding an active form of the enzyme [44,48].

Differential compartmentalization of enzyme and substrate

For the browning process to occur, enzymes have to be present in the same tissues and cell compartments as the substrates and co-substrates (O_2 or H_2O_2) [20,48]. In healthy, non-senescent cells, the enzyme and its substrate(s) are distributed in different subcellular compartments or in different (but adjacent) tissues. Therefore, the oxidation reactions only occur after senescence or an environmental stress (such as pathogen attack or injury) has disorganized the cell or the tissues, and initiated decompartmentalization (i.e. the destruction of the biological barriers between the enzymes and the substrates) [20,48,49]. Vacuoles are common sites for sequestration of anthocyanins, flavan-3-ol monomers, PAs and glycosylated flavonols [3,50]. During senescence, PA polymers move to the cell wall but it is not yet clear how this happens [51].

COs are generally plastid-localized enzymes [44] (Figure 2b), with the exception of aureusidin synthase, which is vacuolar [52]. Most laccases are thought to be secreted proteins [40]. The targeting of these enzymes is conferred by an N-terminal transit peptide (Figure 2b), which is eventually cleaved by a protease, releasing the mature enzyme in the apoplast. PODs are classified according to their subcellular localization, class-I POD being intracellular, and class-III POD being secreted to the apoplast after glycosylation [38,53] (Figure 2b). Interestingly, treatment with methyl-jasmonate has been shown to enhance import and maturation of tomato CO in the chloroplast [54]. This suggests that enzyme compartmentalization is a level of control that is accessible to environmental factors. The dark-brown color in the inner part of a black walnut tree correlates with the presence of PODs and PAs [49,55]. Enzyme and substrate are localized in different cells within the living woody tissue. Therefore, PODs and PAs might interact after they have been released from their respective cell types during the progressive formation of heartwood from sapwood. In Arabidopsis, brown seed coat pigmentation occurs during seed desiccation, when PAs react with the TT10 laccase at the wall of dying testa cells [26] (Figure 4).

Biochemical properties of enzymes

The affinity of PPOs for their phenolic substrates depends on the stereochemistry of the substrate, together with the position and nature of its substitutions (e.g. hydroxyl, methyl and glycosyl) [48]. In contrast to COs, which can only react with *o*-diphenols, laccases can oxidize both *o*and *p*-diphenols, (Figure 2a). Moreover, laccases demonstrate a higher affinity for molecules with a high steric environment than COs do. Plant laccases generally have a low specificity with regard to the reducing substrate [32]. In strawberry, brown polymers resulting from (+)-catechin oxidation are formed mainly by PPOs; PODs appear to be inhibited by the reaction products [56].

Intracellular pH can also be an important factor for these enzyme activities [21,48,57]. For instance, changes in pH values during *in vitro* activities have been shown to prevent proper conformation of the active site, binding of the substrates, and/or catalysis of the reaction [48]. The oxidation of (+)-catechin in the presence of a crude grape PPO extract generates different products depending on whether the reaction occurs at pH 3 or at pH 6 [21].

Physiological roles for flavonoid oxidation

Evidence is accumulating to support the idea that flavonoid oxidation has a protection function during plant development and growth. For example, browning reactions are observed during programmed developmental events such as seed desiccation and plant senescence. Flavonoid oxidation also plays a role in defending the plant against various biotic and abiotic stresses [10,58,59] (Figure 5). Both situations are illustrated below.

Oxidations as part of normal seed and plant development

The *Arabidopsis* TT10 laccase is present in young, colorless seed coats. During the desiccation phase, oxidation of epicatechin and soluble PAs by TT10 might increase their capacity to bind to the cell wall to form, preventively, a physico-chemical protection against stresses. The activity of the TT10 promoter appears to be

strongly induced in early aborted seeds, suggesting that the *TT10* gene might be transcriptionally induced by cell death [26]. During desiccation of pea, cotton or Sida spinosa seeds, flavonoids accumulated in seed coats are oxidized in the presence of PPOs or PODs, leading to seed coat browning and water-impermeability. The formation of quinones and insoluble polymers would explain the reinforcement of the seed coat barrier to water permeation [60–62]. A positive correlation exists between the oxidation of PAs and their cross-linking to the cell wall [8,63]. More generally, such a modification of testa characteristics might reinforce coat-imposed dormancy by increasing the physical resistance of the tissues, impeding the entrance of germination-stimulating agents such as oxygen or water, or the leakage of inhibiting substances such as abscisic acid or carbon dioxide [64-67]. Moreover, the scavenging of oxygen and hydrogen peroxide (H_2O_2) through flavonoid oxidation might protect seeds from deterioration and therefore prolong the period for which they can be stored [65,68].

Browning of onion scales during aging is caused by the autoxidation of quercetin glucosides, after their deglucosylation [69]. Quercetin autoxidation generates activated oxygen (O_2^-) , which reacts with H₂O. This reaction would produce enough H₂O₂ for a POD to enhance quercetin oxidation and lead to the formation of the antifungal agent 3,4-dihydroxybenzoic acid. Finally, oxidation of exogenous flavonoids such as (–)-catechin by root-secreted laccases might play a role in allelopathic interactions between plants [70,71].

Oxidations induced by environmental stresses

The induced production of antimicrobial quinones might lead to the formation of polymeric compounds creating a protective physicochemical barrier [11,61]. Moreover, semiquinones can act as antioxidants by reacting with ROS that appear after tissue damage. Efficient



Figure 5. Flavonoid oxidation: impacts on plant physiology and mechanisms of action. The browning caused by flavonoid oxidation is initiated by developmental (seed desiccation, senescence) and environmental stresses. The main mechanisms of this plant defense reaction are highlighted in gray.

scavenging of ROS has been shown to reduce UV radiation stress [72]. In plant cells, the photosynthetic electron transport system is the major source of ROS such as superoxide (O_2^-) , hydrogen peroxide (H_2O_2) , hydroxyl radical (OH^{\bullet}) , singlet oxygen $({}^1O_2)$, or perhydroxyl radical (HO_2^-) . Flavonoid–POD reaction might function as a mechanism for H_2O_2 -scavenging, and therefore plant cell detoxification [73].

Quinones can also behave as direct toxic compounds against pathogens [53,74]. In Brassica oleracea, POD activity was detected in the integumentary pigment layer, the cotyledons and the embryo axis during imbibition of seeds [75]. Activity peaked after rupture of the integuments, suggesting that guinones have a role in protecting the germinating embryo. In ginseng, reddish-brown areas develop at the root surface as a defense reaction against fungal attack [76]. They have been shown to accumulate phenolic compounds, particularly catechins, and to exhibit enhanced PPO and POD activity. In leaves of drought-stressed tea plants, the formation of epicatechin guinones negatively correlates with lipid peroxidation [77]. Wounding triggers a shift in plant metabolism towards increased biosynthesis of phenolic compounds and enzyme activation, thus providing more antioxidant and antimicrobial agents [20,78]. CO gene expression in hybrid poplar has been shown to be woundand herbivore-induced in young leaves, suggesting that the enzyme might play a role in poplar defense by oxidizing PAs [78]. PA oxidation has been proposed to constitute an antinutritive barrier against herbivores and to protect plant polymers (lignin and cellulose) from degradation by forming complexes with them [11]. Oxidized tannins might also form covalent linkages and interfere with fungal enzymes that are necessary for plant cell invasion (e.g. pectinase, cellulase and laccase). In tomato, two of the seven CO genes are transcriptionally upregulated in abscission zones of leaf petioles in response to water stress [43]. According to the authors, this induction would generate cytotoxic guinones that facilitate cell death during abscission of old leaves. Finally, overexpression and antisense downregulation of a tomato PPO result in enhanced disease resistance and susceptibility, respectively [79,80].

Concluding remarks

Flavonoid oxidation is a complex process involving different mechanisms. Additional research is needed to understand fully the biological functions of oxidative browning in plants. Molecular genetics and the characterization of mutants affected in tissue browning should be useful to confirm and further analyze the physiological roles of flavonoid oxidation products. Finally, further research on the regulation of flavonoid oxidase biosynthesis and activity will be important to elucidate the precise functions and mechanisms of action of these enzymes, including the search for regulatory factors that induce or repress flavonoid oxidation in plants.

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